Future Materials: Titanium, Rare Metals

T. OKABE LAB.

Future Materials: Titanium, Rare Metals

Research Center for Sustainable Material Energy Integration

Resource Recovery and Materials Process Engineering

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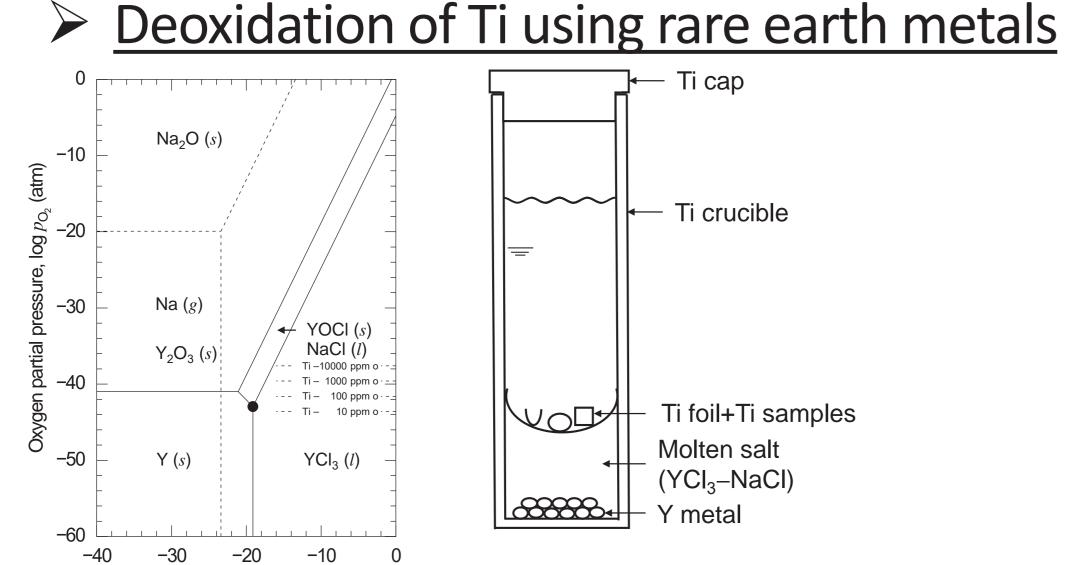
Turning rare metals into "common metals"!!

Okabe Laboratory is working on the research and development of new smelting processes and environmentally friendly recycling processes for rare metals. Our main focus is "future materials: titanium, rare metals". We aim to innovate rare metal process technology for the betterment of society.

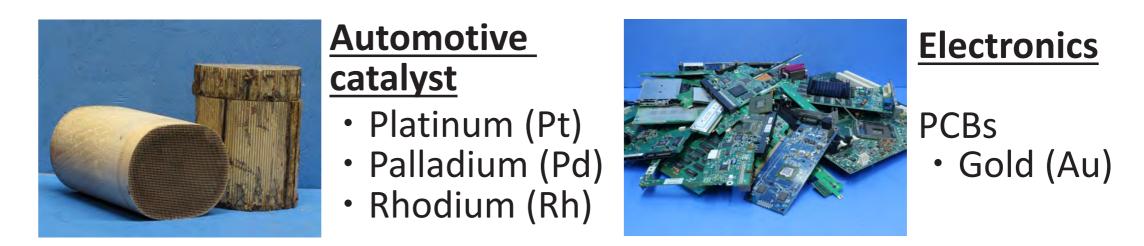


Upgrade recycling of Ti Ti ingot Aerospace Ti scrap such About 14 wt% of a Boeing 787 consists \rightarrow High specific Fabrication as turning of Ti. strength Ti product Marine Structures \rightarrow Corrosion-resistant materials 1000 t of Ti is used for the D runway of the Haneda Airport.

Titanium (Ti) scrap generated in the Ti ingot fabrication process is contaminated by oxygen impurities. Therefore, it is difficult to recycle Ti scraps into ingots.

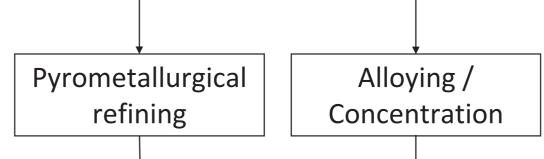


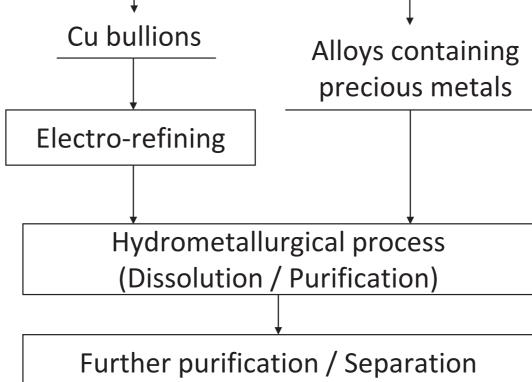
Recycling of precious metals



Automotive catalysts contain 100–1000 times more PGMs than PGM ore. PCBs contain around 10 times more Au than Au ore.

Scraps containing precious metals





To dissolve precious metals into solution, the process

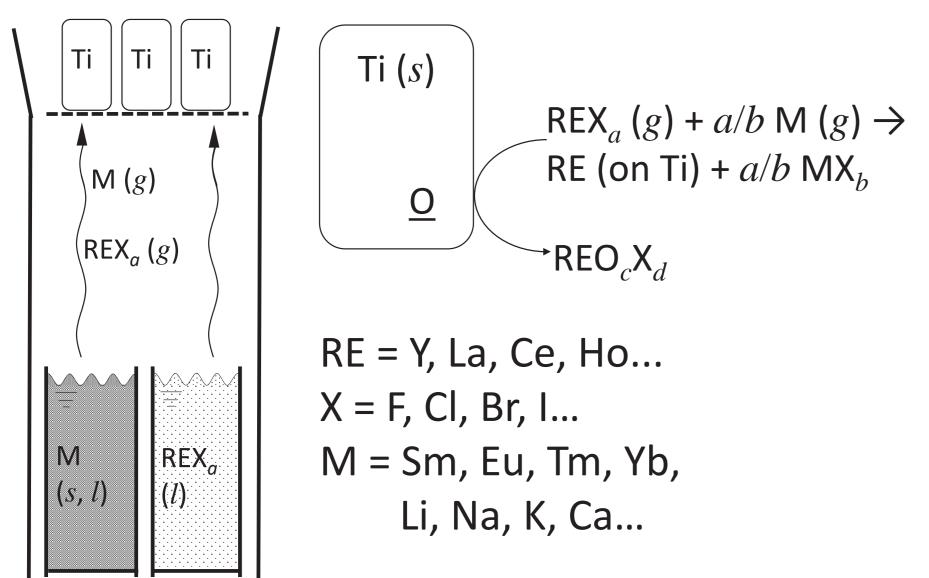
- requires strong chemicals
- generates a large amount of hazardous liquid waste consists of multiple steps



Chlorine partial pressure, $\log p_{Cl_2}$ (atm)

Based on thermodynamic calculations, we have developed a novel deoxidation technology of directly removing oxygen from Ti scrap using rare earth metals such as yttrium (Y) and cerium (Ce).

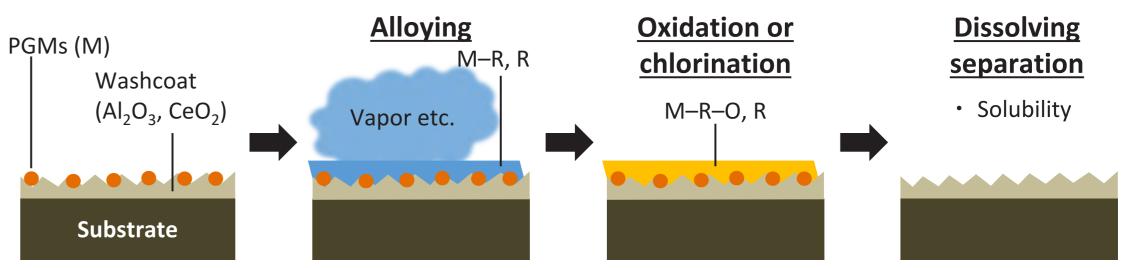
Deoxidation of Ti via gas phase



We have developed a novel deoxidation technology for reducing the amount of deposits on Ti surface and impurities in Ti by supplying deoxidants and fluxes to Ti through the gas phase.

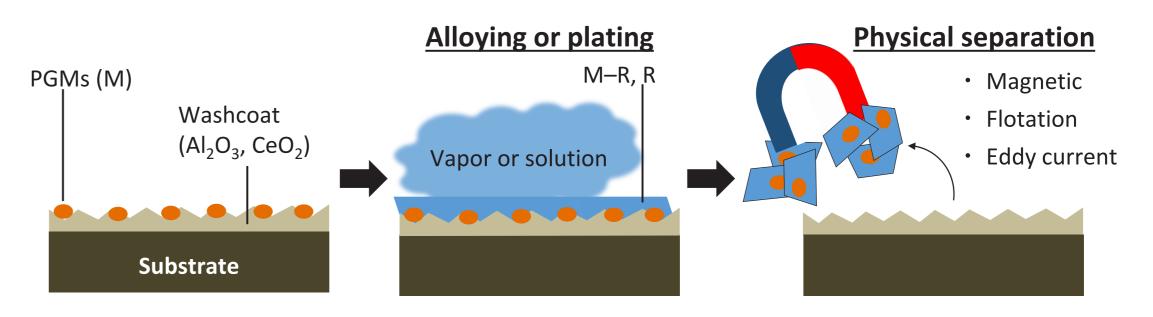
Precious metals

Solubilization process



Technology to dissolve precious metals in non-oxidizing acids (e.g., hydrochloric acid) and solutions (e.g., sea water) using a pretreatment of alloying them with reactive metals such as magnesium (Mg) or calcium (Ca) followed by oxidation or chlorination.

Physical concentration process



Technology to selectively add functions (e.g., magnetism, hydrophobicity) to areas of scrap containing precious metals by electroless plating or chemical treatment with chloride or sulfide vapors, allowing for the concentration of precious metals using physical separation.

